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Pressure study of molecular magnet based on 3d and 4d metals

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The pressure measurements of magnetic materials give the opportunity to study the fundamental aspects of magnetism. In this study a molecular ferromagnet $\text{Co}^{\text{II}}(\text{pyrazol})_4\text{[Nb}^{\text{IV}}(\text{CN})_6]$ has been studied with the use of ac/dc magnetometry under hydrostatic pressure up to 13 kbar. The studied compound crystallizes in the $I4_1/a$ space group where cyanido-bridged structure is decorated with pyrazole molecules coordinated to Co^{II} centers [1]. It is a unique structure with one type of $\text{Co}^{\text{II}}-\text{NC}-\text{Nb}^{\text{IV}}$ linkage. The spin values of both magnetic ions are $\frac{1}{2}$, with $g_{\text{Co}} \approx 4.55$, $g_{\text{Nb}} \approx 2.0$ for cobalt and niobium respectively. The phase transition in ambient pressure occurs at $T_{\text{C}} = 5.4$ K. Initially applying pressure shifts the temperature of phase transition to lower values down to a critical point, after which further applied pressure starts to increase the T_{C} . It has been proved that the compound is changing from a ferromagnetic ordered magnet without pressure to a ferrimagnet in high pressure.

[1] D. Pinkowicz, R. Pełka, O. Drath, W. Nitek, M. Bałanda, A. M. Majcher, G. Poneti and B. Sieklucka, *Inorg. Chem.*, 2010, 49, 7565-7576.

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